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APPLICATION NO.	FIL	ING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO
10/783,729	02/20/2004		Marie S. Chan	5719	6615
7	590	07/29/2005		EXAMINER	
Milliken & Company				HARDEE, JOHN R	
P.O. Box 1927 Spartanburg, SC 29304				ART UNIT	PAPER NUMBER
-F8, -			· ·	1751	

DATE MAILED: 07/29/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)						
Office Action Commons	10/783,729	CHAN ET AL.						
Office Action Summary	Examiner	Art Unit						
	John R. Hardee	1751						
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address						
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If the period for reply specified above is less than thirty (30) days, a reply - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	86(a). In no event, however, may a reply be time within the statutory minimum of thirty (30) days rill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	nely filed s will be considered timely. the mailing date of this communication. D (35 U.S.C. § 133).						
Status								
1) Responsive to communication(s) filed on	_·							
2a) This action is FINAL . 2b) ☐ This	action is non-final.							
	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is							
closed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	53 O.G. 213.						
Disposition of Claims								
4)⊠ Claim(s) <u>1-25 and 27-39</u> is/are pending in the a	application.							
4a) Of the above claim(s) is/are withdraw								
5) Claim(s) is/are allowed.								
6)⊠ Claim(s) <u>1-25 and 27-39</u> is/are rejected.	☑ Claim(s) <u>1-25 and 27-39</u> is/are rejected.							
7) Claim(s) is/are objected to								
8) Claim(s) are subject to restriction and/or	election requirement.							
Application Papers								
9) The specification is objected to by the Examine	r.							
10) ☐ The drawing(s) filed on is/are: a) ☐ acce	epted or b) objected to by the E	Examiner.						
Applicant may not request that any objection to the	drawing(s) be held in abeyance. See	e 37 CFR 1.85(a).						
Replacement drawing sheet(s) including the correcti	ion is required if the drawing(s) is obj	ected to. See 37 CFR 1.121(d).						
11) The oath or declaration is objected to by the Ex	aminer. Note the attached Office	Action or form PTO-152.						
Priority under 35 U.S.C. § 119								
12) ☐ Acknowledgment is made of a claim for foreign a) ☐ All b) ☐ Some * c) ☐ None of:	priority under 35 U.S.C. § 119(a)	-(d) or (f).						
 Certified copies of the priority documents 	s have been received.							
2. Certified copies of the priority documents	• •							
3. Copies of the certified copies of the prior	•	ed in this National Stage						
application from the International Bureau		_						
* See the attached detailed Office action for a list	or the certified copies not receive	a.						
Attachment(s) 1) Notice of References Cited (PTO-892)	A) 🔲 Intonious Successor	(DTO 412)						
2) Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Review (PTO-948)	4) Interview Summary Paper No(s)/Mail Da	nte						
3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08)	. 	atent Application (PTO-152)						
Paper No(s)/Mail Date	6) [_] Other:	\mathcal{O}						

the

DETAILED ACTION

1. Cancelled claim 25 (second instance) and claims 26-38 have been renumbered under Rule 1.26 as claims 26-39, respectively.

Claim Objections

2. Claim 38 is objected to because of the following informalities: This claim is listed as previously presented, but it is new. Appropriate correction is required.

Claim Rejections - 35 USC § 112

- 3. The following is a quotation of the second paragraph of 35 U.S.C. 112:

 The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.
- 4. Claims 1-25 and 27-39 are rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention. What does applicant mean by "less than about 75 parts by weight..."? It appears that the absorbent particulate must be present, but "less than" reads on zero. Is this a mandatory ingredient? If so, applicant should recite a power limit for which basis exists in the specification. If not, it should be indicated as optional.

Claim Rejections - 35 USC § 103

5. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

6. Claims 1-11, 13-16, 21-25 and 27-39 are rejected under 35 U.S.C. 103(a) as being unpatentable over Trinh et al (US 4,481,126).

Trinh discloses a substantially nonabrasive, liquid car cleaner composition which cleans car surfaces without an external source of water to wash or rinse. The product is a composition of up to 30% polymeric solids, up to 95% liquid carrier and a suspension aid. (abstract) Other optional ingredients such as waxes, fluorosurfactants, anticorrosion agents, antistatic agents, sunscreening agents, inorganic mild abrasives, pigments, perfumes, and preservatives can also be used for added benefits. (col. 2, lines 64-68) The liquid car cleaner composition of this invention comprises organic polymeric solids selected from the group consisting of: porous and/or nonporous powdered particles in the particle size range of from 1 micron to about 250 microns (col. 2, lines 37-42) A liquid carrier is required and can be used at a level of up to 95% by weight of the composition. Water and aliphatic hydrocarbon (oil) solvents are used as the liquid carrier. The hydrocarbons can boil as high as 300 degrees C, making obvious the use of mineral oil. Mixtures of water and aliphatic hydrocarbon solvents are preferred. Both surfactants and thickeners are used as the suspending agent. The surfactants are also used as emulsifier and cleaning aid. (col. 2, lines 53-56 and 59-62) The suitable polymeric particulate materials can be synthetic or naturally-occurring polymeric materials include, but are not limited to, polyethylene, polypropylene, polystyrene, polyester resin, urea-formaldehyde resin, polyvinyl chloride, polyacrylics, polyamide, and copolymers, whereas the naturally-occurring polymeric materials are cellulosic materials. (col. 3, lines 34-44) The suspending agents useful in this invention

are suitable surfactants and thickeners and mixtures thereof. These surfactant suspending agents have the properties of dispersing solid particles and liquid droplets. They are used to disperse the polymeric particles throughout the cleaner compositions. Most of the cleaning compositions of this invention contain both oil and water phases. The surfactants also stabilize the emulsion of these two phases. Substantially any surfactant materials which are compatible with the other components in the composition of this invention can be utilized. These include nonionic, anionic, cationic, amphoteric and zwitterionic surfactants. Regarding claim 11, the reference discloses at col. 7, line 13 that nonionic surfactants generally are useful in the compositions, and the structure in claim 11 is generic to most nonionic surfactants. The composition of this invention can consist of up to 10% by weight of a suspending agent surfactant; preferably between 0.4% and 2%. Thickener suspending agents that can be utilized include, but are not limited to, salts of polyacrylic acid polymer, sodium carboxymethyl cellulose, hydroxyethyl cellulose, acrylic ester polymer, polyacrylamide, polyethylene oxide, natural polysaccharides such as gums, algins, pectins. They are used at effective levels of up to 10%. (col. 5, lines 18-45) Although the reference does not teach that the surfactant provides a surface tension in water of about 40 dynes per cm, the compositions are identical and thus the property would be inherent. Example 1 discloses all of the instantly claimed components in their required amounts. (col. 10, lines 46-68)

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The reference fails to teach the particle size of the calcium carbonate used.

The reference fails to teach the specific surfactant of claim 11. The reference fails to teach that the acrylic component that is disclosed is an acrylic stain resistant agent.

Although the reference fails to teach the particle size of the calcium carbonate used, the reference does teach mild inorganic abrasives such as calcium carbonate powder can also be used when polishing action is desired so long as they do not leave unsightly residue on textured vinyl surfaces, (col. 6, lines 28-31) as well as the particle sizes of the other solids that are present, therefore there would be a reasonable expectation of success to modify the prior art to arrive at the instantly claimed invention because the prior art suggest a particle size of other solids to be suspended. Although the reference fails to teach the specific surfactant of claim 11, however there would be a reasonable expectation of success to modify the prior art to arrive at the instantly claimed invention because the prior art does suggest that any surfactant that is compatible with the system may be used. Although the reference fails to teach that the acrylic component that is disclosed is an acrylic stain resistant agent the reference does teach that acrylic additives may be used, therefore there would be a reasonable expectation that material of the same structure will have similar properties.

Thus, it would have been obvious to one of ordinary skill in the art at the time the invention was made to create the instantly claimed composition in view of the Trinh cleaning composition, which contains all the required components in the required amounts.

7. Claims 19 and 20 are rejected under 35 U.S.C. 103(a) as being unpatentable over Trinh et al (US 4,481,126), as relied upon in the rejection above, further in view of Froehlich (US 3910848) or Brown (US 5514302).

The primary reference fails to teach that aerosol may be used with the liquid cleaner of the variety disclosed.

Although the reference does not disclose the use of an aerosol propellant, the use of aerosols with cleaning compositions is well known in the art. Froehlich, the secondary reference in analogous art teaches that a cleaning composition containing a polymer urea-formaldehyde polymer particles having a particle size of from 10 to 105 microns and an oil value of at least 90, a halogenated solvent boiling at from 45 degrees. to 120 degree C., a silica antisettling agent, a cationic antistatic agent, and an aerosol propellant selected from at least one of trichlorofluoromethane, dichlorodifluoromethane, 1,2-dichlorotetrafluoroethane, propane, isobutane and butane. (col. 1, lines 37-60).

Therefore there is a reasonable expectation of success that an aerosol may be used with the composition of the reference as the composition of the secondary reference has similar structural properties, uses and components.

Brown, the secondary reference in analogous art teaches an improved aqueous fabric cleaning shampoo composition fabric solid cleaning polymer, surfactant in water Which may be in the form of a self-pressurized aerosol, with a conventional propellant such as dimethyl ether or one or more saturated alkanes containing from 2 to 6 carbon atoms such as propane, isopropane, n-butane, isobutane, isopentane or n-hexane is

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added through the valve. Although the reference does not disclose the use of an aerosol propellant, the use of aerosols with cleaning compositions is well known in the art. Brown, (abstract col. 10, lines 27-48).

Therefore there is a reasonable expectation of success that an aerosol may be used with the composition of the reference as the composition of the secondary reference has similar structural properties, uses and components.

8. Claims 1-4, 6-15, 17, 18 and 27-38 are rejected under 35 U.S.C. 103(a) as being unpatentable over Suzuki et al (US 4,534,892). Suzuki discloses a liquid detergent composition containing an anionic surface active agent and a water-insoluble fine powder, characterized by containing therein a cross linking type amphoteric polymer and an inorganic salt. (abstract) The reference teaches that several glycols and phosphates may be used and that the foaming properties are improved using a polyacrylic acid and as a pH adjuster. (col. 3, lines 15-25 and col 1. lines 48 and 54-63). Inorganic salts used in the present invention are alkali metals, alkaline earth metals or aluminum salts of hydrochloric acid, sulfuric acid, nitric acid, etc. Of such inorganic salts, preferred are potassium sulfate, sodium sulfate, magnesium sulfate, aluminum sulfate, potassium nitrate, sodium nitrate, magnesium nitrate, calcium nitrate, aluminum nitrate, potassium chloride, sodium chloride, magnesium chloride, calcium chloride, aluminum chloride, potassium carbonate, sodium carbonate, and aluminum carbonate, and particularly sodium sulfate, potassium nitrate, sodium nitrate, potassium chloride and sodium chloride are preferable. With respect to the amount of the inorganic

salt to be incorporated into the liquid detergent composition of the present invention, the ratio (weight ratio) of the inorganic salt relative to the cross linking type amphoteric polymer is important, and the amount of the inorganic salt should be 10 to 1/10 times that of the cross linking type amphoteric polymer. (col. 5, lines 25-44) Alkyl- or alkenylethoxy sulfates having straight or branched chain alkyl or alkenyl groups with the average carbon number of 8-20 and having added ethylene oxide of 0.5-8 moles on average to the molecule. (col. 5, lines 56-59) Furthermore, although water-insoluble fine powder used in the present invention is not particularly limited to specific ones, preferably usable are pigments such as silicon dioxide, aluminum oxide, magnesium oxide, titanium oxide, aluminosilicate, silicon carbide, calcium carbonate, calcium phosphate, chromium oxide, barium carbonate, Hansa Yellow, talc, etc.; pearling agents or clouding agents such as mica, fish scale, etc. and germicidal preservatives as zinc-2pyridyl-thio-1,1'-dioxide; The fine powder usually has a particle size of less than 150 microns. (col. 6, lines 42-58) Nonionic surface active agents, amphoteric surface active agents and cationic surface active agents may be used in combination with the anionic surface active agents. Other components which may be incorporated are dissolving agents such as propylene glycol, glycerin, urea, etc.; viscosity regulators such as ethanol, isopropanol, higher alcohol, hydroxyethyl cellulose, hydroxypropyl cellulose, etc.; perfumes, dyes, ultraviolet absorbers, antioxidants, water-soluble anti-dandruff agents, sterilizers, preservatives, etc. The liquid detergent composition of the present invention obtained in the aforementioned manner is applicable to various purposes such

as detergent for wool and silk, body shampoo, detergent for fine fabric, etc. (col. 7.

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lines 7-29) Example 2 teaches the use of 2-methacryloxyethyl diethylammonium ethosulfate (col. 9, lines 39-58) and all the example discloses that water makes up the balance of the compositions.

Choice of one of any common biocides would be obvious in the absence of unexpected results.

Response to Arguments

9. Applicant's arguments filed April 12, 2005 have been fully considered but they are not persuasive. Applicant argues that the recitation of "consisting essentially" scope distinguishes he invention over Trinh et al. This is not persuasive because the silicone does not "materially affect" the properties of the cited compositions. They are cleaning compositions, and applicant's claimed compositions are cleaning compositions... Applicant argues that a vacuum retrieval material is not disclosed in the reference. Polyethylene oxide is specifically mentioned in the rejection, regardless of whether or not it is called a vacuum retrieval agent, so it meets the limitation. Applicant argues that an acrylic stain resist agent is not disclosed. This is not persuasive because the reference teaches the use of acrylates. Applicant's recitation of a stain resist agent amounts to a recitation of intended use. Applicant argues that the Suzuki reference does not teach the use of a dispersion stabilizing agent. This is not persuasive because the reference teaches the use of acrylic copolymers which can comprise acrylic acid and acrylic esters such as dimethylamino ethyl methacrylate. It is clear from applicant's specification that acrylic acid/acrylic ester copolymers fall within the ambit of applicant's

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claims. Regarding claim 11, the reference discloses at col. 7, line 13 that nonionic surfactants generally are useful in the compositions, and the structure in claim 11 is generic to most nonionic surfactants.

- 10. This action contains grounds of rejection not motivated by applicant's amendments. Accordingly, it is NOT FINAL.
- 11. Any inquiry concerning this communication or earlier communications from the examiner should be directed to the examiner, Dr. John R. Hardee, whose telephone number is (571) 272-1318. The examiner can normally be reached on Monday through Friday from 8:00 until 4:30. In the event that the examiner is not available, his supervisor, Dr. Yogendra Gupta, may be reached at (571) 272-1316.

The fax phone number for the organization where this application or proceeding is assigned is (571) 273-8100.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR.

Status information for unpublished applications is available through Private PAIR only.

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John R. Hardee Primary Examiner

July 26, 2005